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Responses of wheat (*Triticum aestivum* **L.)** genotypes to sulfur aerosols

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Abstract

An investigation into the effects of S- aerosols on five wheat genotypes (viz., GW-322, GW-366, GW-273, GW-173, JW-336), was carried out in field (October, 2016-March, 2017) at Assam Agricultural University (ICR Farm). Simulated S-aerosols viz., (NH₄)₂SO₄, CaSO₄, and K₂SO₄: @ 300 ppm each (\approx 30 kg N ha⁻¹) along with a control were misted on the plants, on sunny days in the afternoon (after 2–3 P.M.) at three different growth stages i.e. seedling, maximum tillering and spike initiation stages. Therefore, the total concentration of each of the S-aerosols was 900 ppm \approx 0.9%. The S-aerosols affected the wheat varieties variably which was proved by the significant changes of net photosynthesis rate (Pⁿ), chlorophyll accumulation, enzyme nitrate reductase activity, cell membrane stability (CMS) and cellular [Ca²⁺] and [K⁺] concentrations at maximum tillering and spike initiation stages of the crop. The S and N contents in grains, nitrogen use efficiency (NUE) and sulfur use efficiency (SUE) were also analysed for the crop at harvest. The S-aerosols could augment the SUE and NUE in the wheat crop. The genotype GW-366 was the most responsive under the influence of foliar fertilization with S-aerosols. Among the S-aerosols, (NH₄)₂SO₄ was the most effective in the work.

Keywords: wheat, S-aerosols, calcium, potassium, photosynthesis and NUE, SUE, CMS, lipid peroxidation

Introduction

Wheat is the second most important cereal crop next to rice in India. Wheat occupies an area of 24.23 million hectares with a production of 70.26 million tonnes in the country (Annonymous, 2011-12)^[1]. Sulfur plays important role in wheat productivity as it is the constituent of several amino acids viz., cysteine, methionine, thioredoxins, sulfolipds and coenzymes such as biotin, coenzyme-A, thiamine pyrophosphate and lipoic acid, (Ernst, 1993)^[9]. In the past, the responses of crop plants to basal Sulfur have been studied (Singh *et al.*, 2014)^[24]. But, little information is available on the physiological responses of wheat crop to Sulfur while applied in the form of aerosols. The responses of wheat crop are explored in the present investigation.

Materials and methods

Experimental site and situation

The investigation was carried out at the ICR farm, Assam Agricultural University, Jorhat during the year 2016-17. Jorhat is situated at 26°45' N latitude and 94°12'E longitude with an altitude of 87 meters above mean sea level. The climatic conditions of Jorhat as a whole, is subtropical, humid, dry summer and cold winter.

The aerosol treatment condition

The crop was treated with aerosols, and cultivated in the meteorological conditions of cold winter (9.4-29.78°C) with high humidity (55-99%), low rainfall (0.15-3.78mm) and lower bright sunshine (4.16-6.24) hours (Table 1).

Experimental materials

Five wheat varieties (*viz.*, GW-322, GW-366, GW-273, GW-173, JW-336) were collected from the eastern wheat-growing zone of India (*viz.*, Uttar Pradesh), and used in the experiment.

Crop husbandry

Seeds were put in a container, and Captan @ 2.5g kg⁻¹ seed was added to it. The fungicide was mixed thoroughly with seeds by agitating them for five minutes. The plots were ploughed

thoroughly, mowed and leveled. Recommended doses of N,P,K fertilizers @ 80:46:42 per hectare were applied as basal. The Randomised Block Design (RBD) with two replications was followed in the experiment. The crop was

irrigated regularly during the growth period. The plots were kept weed free always manually. Prophylactic measures were taken to prevent the crop from the attack of insects and pests.

 Table 1: Meteorological data during the crop season (December, 2016 to March, 2017)

Months	Tempera	ture (°C)	Average Relative Humidity	Average Relative Humidity	Monthly total	Monthly total Bright
wontins	Max.	Min.	in the morning (%)	in the evening (%)	Rainfall (mm)	sunshine (hours)
December	26.3	11.9	99	62	0.15	6.24
January	25.2	9.4	98	57	2.1	5.77
February	26.8	13.0	95	55	1.38	4.99
March	29.78	15.74	97	58	3.78	4.16
Total					7.41	21.16

Misting of aerosols on plants

The foliages of plants were misted with S-aerosols @ 300 ppm (\approx 30 kg S ha⁻¹) at three growth stages of the crop *viz.*, at seedling stage, maximum tillering stage and spike initiation stages. Each of the S-aerosol (1000ml for a single stage) was applied in 3 splits on cloud free and clear sunny days in the afternoon when air temperature was low. The cumulative volume of one aerosol solution was 0.9% (\approx 300 ppm x 3=900ppm) only. While spraying the aerosol solutions, its drifting was checked from one plant to another using hard board as partition between two plots. A digital pH meter with standard pH (4&7) was used to measure the pH of the S-aerosols which were found as distilled water:7.00; (NH₄)₂SO₄: 5.34; CaSO₄: 5.49, K₂SO₄:5.66

Net photosynthesis (P^n) measurement

In the experiment, plants were incubated, and P^n was measured using a transparent, airtight acrylic assimilation chamber (volume 15x15x15 cm³). Leaves of uniform small size (5-15nos.) were incubated in a condition of ambient (400ppm) CO₂, temperature (22⁰C) and light intensity (22.56 µm photons m⁻² sec⁻¹) in the laboratory. A volume of 10 cm³ air samples was collected by clinical syringe injecting through the rubber port of the chamber, and it was pushed into the Environmental Gas Monitor (EGM-4) through its port. Carbon dioxide concentration after incubation was recorded as displayed by the EGM. Then, ppm-CO₂ absorbed per gram plant dry weight per hour was expressed as the rate of net photosynthesis (Larsson and Kershaw, 1975)^[15].

Determination of Chlorophyll content in leaves

A standard protocol as suggested by Arnon (1949)^[2] was used to estimate the total chlorophyll content of the leaves. Ten uniform fresh leaf discs of 5 mm diameter were cut, weighed, and put into 10 ml of 80% acetone in pyrax glass tube. The test tube was sealed by polythene and rubber band to prevent loss of acetone by evaporation. Aluminum foil was used to cover the test tubes for protecting the chlorophyll degradation by photo oxidation. Then, the set up was refrigerated at 4°C for 72 hours. The, supernatant was decanted, its final volume was recorded, and its spectrometric readings were recorded using wave band 645 nm and 663 nm.

Revealing of Nitrate Reductase (NR) activity in plants

Fresh green leaves were collected from field in ice polythene bags for estimation of in- vivo NR activity (Keeper *et al.*, 1971) ^[14]. Leaf samples of 10-15 mm² size (200mg) were cut and put into 2.5 ml solution of 200 mM phosphate buffer (pH 7.5), 30 mM KNO₃, 5 % (v/v) propanol in assay tubes. The samples were incubated for 30 minutes at $33\pm2^{\circ}$ C. Then, the tubes were put in boiling water bath for 15 minutes to stop the

reaction of the mixture. After cooling, 0.2 ml of the reaction mixture was taken into a test tube to which 1 ml of 1% sulphanilamide and 1 ml of 0.025 N-(1-napthyl)-ethylene diamine dihydrochloride solution were added, and kept for 15 minutes to develop the pink colour in the solution. Finally, the absorbance of the solution was read at 540nm in spectrophotometer. A standard curve for nitrite assay was prepared for calculation of NR activity which was expressed as μ moles NO₂ formed g⁻¹fresh tissue wt. hr⁻¹.

Assessment of Cell membrane stability (CMS) and detection of cellular $[Ca^{2+}] \& [K^+]$

Fresh leaf samples were collected in polythene ice bags. Twenty pieces of leaves (1cm² size) were immersed into 20 cm³ distilled water in plastic bottles of 60 cm³ capacity. The bottles were made air tight to avoid leaking of the solution. The samples were checked using magnetic stirrer using electrically operated Vortex Mixture. Thus, leaf samples were washed thrice with distilled water (each 10 min, 20 cm³) to collect the intercellular ions. Then, the exchangeable ions of the same leaf tissues were extracted by eluting twice in 25 mM Sr₂Cl. (each 1 h 20 cm³) solution. The solutions were collected in plastic bottles, and the plant samples were oven dried at 60°C to a constant weigh. The CMS was calculated using the electrical conductivity readings of these solutions as per the protocol suggested by Bharali et al. (2015) [4]. The Flame photometric method (Jackson, 1973)^[13] was used to measure the [K⁺], whereas, colorimetric titration method (Richards, 1954) ^[21] was employed for estimation of intercellular and exchangeable [Ca²⁺]. The leaf tissues were oven dried, and ion contents were expressed as mM per gram dry weight of the leaves.

Nitrogen use efficiency (NUE)

The modified Kjeldahl method (Jackson, 1973) ^[13] was used to estimate total Nitrogen (%) in grains based on the catalytic conversion of organic nitrogen into ammonia, and its acid base titration. The grain NUE was calculated as the multiplication of the per cent Nitrogen in grain by total grain yield per unit area.

Sulfur use Efficiency (SUE) in grains

The SUE of the wheat crop was calculated by multiplying the per cent S content of grain with the grain yield (tha⁻¹).

Statistical analysis

Data for each plant parameter was analysed by Fisher's method of analysis of variance (Panse and Sukhatme, 1978)^[19]. Significance or non-significance of variance due to the treatments was determined by the respective 'F' values. The

standard error of the means (S.Ed. $\pm)$ was calculated by using the following expression.

S.Ed(
$$\pm$$
) = $\frac{\sqrt{2 \times \text{error mean square}}}{\text{Pooled number of replication}}$

The critical difference between a pair of treatment means was judged by comparing the values obtained from the product of S.Ed (\pm) and Probability at (0.05).

Results

The results obtained in the investigation into the 'Responses of wheat (*Triticum aestivum* L.) genotypes to Sulfur aerosols' under field conditions are presented in tabular form. The main aim of the work was to study the responses of the wheat genotype in terms of higher SUE, and NUE, to foliar application of Sulfur aerosols. The study, too, delved into the mechanism of altering the physiology of wheat crop by Saerosol.

Data presented in Table 2 showed that there were significant differences in Pⁿ among the varieties, and among treatments at maximum tillering but for treatments only at spike initiation stage. Among the varieties, the GW-322 (6601.4ppm CO₂ absorbed g⁻¹d.w.hr⁻¹) had the highest Pⁿ value followed by GW-273 (6521.1 ppm CO₂ absorbed g⁻¹d.w.hr⁻¹)> GW-366(6459.7 ppm CO₂ absorbed g⁻¹d.w.hr⁻¹)> GW-366(6459.7 ppm CO₂ absorbed g⁻¹d.w.hr⁻¹)> GW-173(6306.26 ppm CO₂ absorbed g⁻¹d.w.hr⁻¹), and the lowest Pⁿ was found in JW-336(5932.5ppm CO₂ absorbed g⁻¹d.w.hr⁻¹). On an average, among the S aerosols, (NH₄)₂SO₄(6827.9ppm CO₂ absorbed g⁻¹d.w.hr⁻¹) had the highest Pⁿ as compared to control (5446.56ppm CO₂ absorbed g⁻¹d.w.hr⁻¹) at maximum tillering stage.

At spike initiation stage, the variety GW-173 (6648.1ppm CO₂ absorbed g⁻¹d.w.hr⁻¹) showed the maximum Pⁿ followed by GW-322 (6535.9 ppm CO₂ absorbed g⁻¹d.w.hr⁻¹)>GW-273 (6448.9 ppm CO₂ absorbed g⁻¹d.w.hr⁻¹)>GW-366 (6116.7ppm CO₂ absorbed g⁻¹d.w.hr⁻¹), and the lowest Pⁿ was recorded in JW-336 (5905.7ppm CO₂ absorbed g⁻¹d.w.hr⁻¹) at the spike initiation stage. On an average, (NH₄)₂SO₄ registered the highest Pⁿ (6501.28 ppm CO₂ absorbed g⁻¹d.w.hr⁻¹) as compared to the control (5788.4 ppm CO₂ absorbed g⁻¹d.w.hr⁻¹).

Data presented in Table 3 reveal that the total chlorophyll content in leaf tissue varied significantly among the treatments, and among the varieties at the maximum tillering and spike initiation stages. The highest total chlorophyll content was found in the variety GW-273(2.093mg g⁻¹fw of leaf) >GW- 336 (2.034mg g⁻¹fw of leaf) >GW-322 (1.977mg g⁻¹fw of leaf) >GW-173 (1.976), and the lowest value was in JW-336 (1.847mg g⁻¹fw of leaf). Among the S-aerosol treatments, (NH₄)₂SO₄ recorded the highest total chlorophyll content in leaf tissue (2.262 mg g⁻¹fw of leaf) as compared to the control plants (1.704mg g⁻¹fw of leaf).

At spike initiation stage, over all, the variety GW-273 recorded the highest total chlorophyll content (2.002mg g⁻¹fw of leaf) >GW-366 (1.962mg g⁻¹fw of leaf) >GW-322 (1.906 mg g⁻¹fw of leaf) >GW- 173 (1.879 mg g⁻¹fw of leaf), and the lowest one was found in JW-366 (1.777mg g⁻¹fw of leaf). On an average, among the S-aerosol treatments, (NH4)₂SO₄ produced the maximum total chlorophyll content in leaf tissues (2.191 mg g⁻¹fw of leaf) as compared to the controlled plants (1.607 mg g⁻¹fw of leaf).

Data presented in Table 4 indicate that there were significant variations in nitrate reductase (NR) activity in leaf tissues among the treatments, and among the varieties at the

maximum tillering stage. The highest NR activity was observed in GW-273 (0.698 nmol NO₂⁻ g⁻¹fw of leaf hr⁻¹) >GW- 366 (0.684nmol NO₂⁻ g⁻¹fw of leaf hr⁻¹) >GW-322 (0.644nmol NO2⁻ g⁻¹fw of leaf hr⁻¹) GW-173 (0.643nmol NO2⁻ g⁻¹fw of leaf hr⁻¹), and the lowest was in JW-336 (0.573 nmol NO₂⁻ g⁻¹fw of leaf hr⁻¹). Among the S-aerosol treatments, (NH₄)₂SO₄ exhibited the highest NR activity (0.705nmol NO₂g⁻¹ fw of leaf hr⁻¹) as compared to the Controlled plants (0.532 nmol NO₂⁻ g⁻¹ f.w. of leaf hr⁻¹) at the maximum tillering stage. The data presented in Table-4 also reveal nonsignificant differences of NR activity among the treatments, and a significant variation of NR activity among the varieties at spike initiation stage. The highest NR activity was in GW-366 (0.719nmol NO₂⁻ g⁻¹fw leaf hr⁻¹) followed by GW- 273 (0.707 nmol NO₂⁻ g⁻¹ fw of leaf hr⁻¹) >GW-322 (0.685nmol NO_2^{-} g⁻¹ fw of leaf hr⁻¹) >GW-173 (0.676 nmol NO_2^{-} g⁻¹ fw of leaf hr-1), and the lowest NR activity was in JW-336 (0.618nmol NO₂⁻ g⁻¹ fw of leaf hr⁻¹). On an average, among the S-aerosol treatments, (NH₄)₂SO₄ activated the highest NR (0.7nmol NO₂⁻ g⁻¹ fw of leaf hr⁻¹) as compared to the controlled one (0.65nmol NO₂⁻ g⁻¹ fw of leaf hr⁻¹) at spike initiation stage.

Data in Table 5 reveal that there were non-significant differences of lipid peroxidase activity among the treatments and among the varieties at the maximum tillering and spike inituiation stages. At maximum tillering stage, the highest lipid peroxidation in leaf was observed in GW- 273 (3.376 nmol MDA g⁻¹ f.w) followed by GW-322 (2.955 nmol MDA g-1 f.w) >GW-366 (2.653 nmol MDA g-1 f.w) >GW-173 (2.359 nmol MDA g⁻¹ f.w⁻), and the lowest was observed in JW-336 (2.157 nmol MDA g⁻¹ f.w). On an average, among the S-aerosol treatments, plants treated with CaSO₄ had the highest rate of lipid peroxidation (2.976nmol MDA g⁻¹ f.w) as compared to controlled plants (2.449nmol MDA g⁻¹ f.w). At spike initiation stage, on an average, the highest lipid peroxidation was found in JW-336 (3.843 nmol MDA g⁻¹ f.w) >GW- 173 (3.752 nmol MDA g⁻¹ f.w), > GW-366 (3.737 nmol MDA $g^{-1}f.w$) >GW-273 (3.689 nmol MDA $g^{-1}f.w$), and the lowest was recorded in GW-322 (3.574 nmol MDA g⁻¹ f.w). The plants treated with CaSO₄ showed the highest lipid peroxidatn (4.108 nmol MDA g⁻¹ f.w) as compared to controlled plants (3.352 nmol MDA g⁻¹ f.w) at spike initiation stage also.

The data displayed in Table-6 indicate non-significant variations in CMS among the genotypes and among the treatments at both the maximum tillering and spike initiation stages. At maximum tillering stage, the highest CMS was exhibited by GW- 322 (0.33)>followed by GW-273 (0.28)>GW-366 (0.269) >JW-336 (0.236), and the lowest CMS was in JW-336 (0.2). On an average, the highest CMS was in CaSO4 (0.262), and the controlled plants (0.295) had the lowest of it at maximum tillering stage. At spike initiation stage, among the varieties, the highest CMS was found in GW-173 (0.3)>GW-273 (0.251)>JW-336 (0.213)>GW-366 (0.15)>GW-173 (0.02), and the lowest CMS was found in JW-322 (0.131). On an average, the maximum CMS found in CaSO4 (0.23) and the lowest was in case of controlled plants (0.015) at spike initiation stage.

Data portrayed in Table 7 (A&B) reveal that sulfur and nitrogen contents in wheat seed at harvest varied non significantly among the treatments and among the genotypes (except for N). On an average, the maximum sulfur content in seed was found in GW-322((0.347)>GW-273 ((0.341%)>GW-173((0.323%)> JW-336((0.303%)), and the lowest sulfur content was found in GW-366 ((0.283%)). Plants treated with

 $(NH_4)_2SO_4$ (0.328%) had the highest value as compared to the controlled plants (0.316%). On an average, the maximum Nitrogen content in seed was found in JW-336 (0.35%)>GW-173(0.293%)>GW-322(0.267%) >GW- 273 (0.238%), and the lowest was found in GW-366 (0.193%). Plants treated with $(NH_4)_2SO_4$ (0.34%) had the highest N contents as compared to control (0.243%).

Data in Table 8 (A) reveal SUE in seed varied significantly among the treatments but it was non-significant among the genotypes. On an average, the maximum Sulfur content in seed was found in GW-366 (0.199%)>GW-322(0.189%) >GW-273 (0.178%)>GW-173(0.172%), and the lowest was in JW-336(0.157%). Among the aerosol treatments, $(NH_4)_2SO_4$ (0.23%) showed the highest value in plants as compared to the control (0.131%). Data presented in Table 8(B) reveal that NUE varied significantly among the treatments and among the genotypes. On an average, the maximum NUE in seed was found GW-322(0.217%)>GW-366(0.216%) >GW-273 (0.211%)>GW-173(0.148%), and the lowest was found in JW-336 (0.125%). Plants treated with (NH₄)₂SO₄ (0.253%) had the highest NUE as compared to control (0.141%).

Data presented in Table 9 reveal that there were nonsignificant variations of intercellular and exchangeable $[Ca^{2+}]$ among the varieties and among the treatments (except intercellular $[Ca^{2+}]$). On an average, the variety GW-366 (1.428 mgg⁻¹d.w.) registered the highest intercellular $[Ca^{2+}]$ followed by JW-336 (1.28 mgg⁻¹d.w.)>GW-273,173 $(1.224 \text{mgg}^{-1} \text{d.w.})$ while the lowest $[\text{Ca}^{2+}]$ was in GW-322 (1.204 mgg⁻¹d.w.). Overall, CaSO₄ showed the maximum $[Ca^{2+}]$ (1.511mgg⁻¹d.w.) as compared to control (0.945mgg⁻¹ ¹d.w). On an average, the variety JW-336(1.316mgg⁻¹d.w.) registered the highest exchangeable [Ca²⁺] followed by GW-322 (1.305mgg⁻¹d.w.)> GW-366(1.286mgg⁻¹d.w.)>GW-173 (1.246mgg⁻¹d.w.), while the lowest was in GW-273 (1.21mgg⁻¹ ¹d.w.). Overall, CaSO₄ showed the maximum $[Ca^{2+}]$ (1.396mgg⁻¹d.w) as compared to Control (1.26mgg⁻¹d.w.). Data presented in Table 10 reveal that there were nonsignificant variations of intercellular and exchangeable [K⁺] among the varieties and among the treatments. On an average, the variety GW-366 (1.909mgg⁻¹d.w.) registered the highest intercellular [K⁺] followed by GW-273 (1.644)> JW-336(1.613) >GW-173(1.499mgg⁻¹d.w.), while the lowest intercellular [K⁺] was in GW-322 (1.306mgg⁻¹d.w.). Overall, K_2SO_4 (1.741mgg⁻¹d.w) produced the maximum intercellular [K⁺] as compared to the control (1.441mgg⁻¹d.w.). The variety GW-366 (2.079mgg⁻¹d.w) registered the highest exchangeable $[K^+]$ followed by GW-173 (1.791mgg⁻¹d.w.)> JW-336(1.648)>GW-273(1.638mgg⁻¹d.w.)>GW-322 (1.458mgg⁻¹d.w.)>GW-322 (1.458mgg⁻¹d.w.))>GW-322 (1.458mgg⁻¹d.w.))</br> ¹d.w), while the lowest $[K^+]$ was in GW-322 (1.889mgg⁻ ¹d.w.). Overall, K_2SO_4 showed the maximum [K⁺] (1.955mgg⁻ ¹d.w.), as compared to the control (1.518mgg⁻¹d.w.).

Table 2: Effect of sulfur aerosols on rate of net photosynthesis (Pⁿ) of wheat crop at maximum tillering and spike initiation stages

	P ⁿ (ppm C	O2 absorbed g	g ⁻¹ d.w hr ⁻¹) a	t the maximu	ım tillering	P ⁿ (ppm CO ₂ absorbed g ⁻¹ d.w hr ⁻¹) at spike initiation					
Variety (V)↓			stage			stage					
Treatment $(T) \rightarrow$		Sulfur	aerosols (30 l	kg ha ⁻¹)		Sulfur aerosols (30 kg ha ⁻¹)					
	CaSO ₄	(NH4)2SO4	K ₂ SO ₄	Control	Mean	CaSO ₄	(NH4)2SO4	K ₂ SO ₄	Control	Mean	
GW-322	6,582.40	6,862.15	7,077.95	5,883.30	6,601.45	6,584.00	6,555.45	6,865.25	5,600.70	6,401.35	
GW-366	6,295.50	7,133.95	6,760.85	5,648.65	6,459.74	6,183.95	6,915.20	5,650.60	5,716.95	6,116.68	
GW-273	6,609.35	7,344.55	6,733.35	5,397.30	6,521.14	6,383.85	6,733.35	7,278.65	5,399.65	6,448.88	
GW- 173	6,584.85	6,683.35	6,732.40	5,224.43	6,306.26	6,616.70	6,583.60	6,708.30	6,683.90	6,648.13	
JW-336	6,492.60	6,115.55	6,042.85	5,079.10	5,932.53	6,545.55	5,718.80	5,817.45	5,541.00	5,905.70	
Mean	6,512.94	6,827.91	6,669.48	5,446.56		6,462.81	6,501.28	6,464.05	5,788.44		
Factor		CD (0.05)	S.Ed(±)			CD (0.05)	S.Ed(±)				
Factor (V)		239.932	113.794			n.s.	250.885				
Factor (T)		214.602	101.781			473.136	224.398				
Interaction (TxV)		479.864	227.589			n.s.	501.769				

Table 3: Effect of Sulfur aerosols on total Chlorophyll (Chll) content of wheat crop at maximum tillering and spike initiation stages

Variety (V)↓	Total Chll (mg g ⁻¹ f.w. o tilleri	f leaf) content ng stage	Total Chll (mg g ⁻¹ f.w. of leaf) content at spike initiation stage							
Treatment $(T) \rightarrow$	Sulfur aeros	sols (30kg ha ⁻	¹)		Sulfur aerosols (30kg ha ⁻¹)					
	CaSO ₄	(NH4)2SO4	K ₂ SO ₄	Control	Mean	CaSO ₄	(NH ₄) ₂ SO ₄	K ₂ SO ₄	Control	Mean
GW-322	1.904	2.36	2.048	1.595	1.977	1.839	2.33	1.957	1.499	1.906
GW-366	2.147	2.44	1.91	1.638	2.034	2.085	2.305	1.841	1.617	1.962
GW-273	1.912	2.499	2.198	1.765	2.093	1.823	2.426	2.093	1.667	2.002
GW- 173	1.799	2.197	2.175	1.735	1.976	1.75	2.125	2.087	1.555	1.879
JW-336	1.717	1.815	2.066	1.79	1.847	1.693	1.769	1.953	1.695	1.777
Mean	1.896	2.262	2.079	1.704		1.838	2.191	1.986	1.607	
Factor	CD (0.05)	S.Ed(±)			CD (0.05)	S.Ed(±)				
Factor (V)	0.042	0.02			0.059	0.028				
Factor (T)	0.037	0.018			0.053	0.025				
Interaction (TXV)	0.084	0.04			0.118	0.056				

Table 4: Effect of Sulfur aerosols on Nitrate reductase activity of wheat crop at maximum tillering and spike initiation stages

	Nitrate re	eductase activity	(nmol NO2 ⁻	g ⁻¹ f.w. of leaf	hr-1) at	Nitrate reductase activity (nmol NO ₂ ⁻ g ⁻¹ f.w. of leaf hr ⁻¹)					
Variety(V)↓		the maxim	um tillering	stage		at spike initiation stages					
Treatment(T)→		Sulfur aer	osols (30kg l	ha ⁻¹)		Sulfur aerosols (30kg ha ⁻¹)					
	CaSO ₄	(NH4)2SO4	K ₂ SO ₄	Control	Mean	CaSO ₄	(NH4)2SO4	K ₂ SO ₄	Control	Mean	
GW-322	0.694	0.717	0.706	0.46	0.644	0.717	0.73	0.717	0.579	0.685	
GW-366	0.712	0.739	0.7	0.587	0.684	0.726	0.75	0.703	0.699	0.719	
GW-273	0.661	0.765	0.72	0.647	0.698	0.667	0.76	0.73	0.67	0.707	
GW- 173	0.662	0.696	0.677	0.537	0.643	0.684	0.658	0.682	0.68	0.676	
JW-336	0.628	0.61	0.622	0.432	0.573	0.593	0.605	0.648	0.625	0.618	
Mean	0.671	0.705	0.685	0.532		0.677	0.7	0.696	0.65		
Factor		CD (0.05)	S.Ed(±)			CD (0.05)	S.Ed(±)				
Factor (V)		0.015	0.007			0.016	0.008				
Factor (T)		0.013	0.006			0.014	0.007				
Interaction	(TxV)	0.029	0.014			0.032	0.015				

Table 5: Effect of Sulfur aerosols on lipid peroxidase activity of the wheat crop at maximum tillering and spike initiation stages

	Lipid peroz	xidation (nmo	ol MDA g ⁻¹ f	.w.) at the	maximum	Lipid peroxidase (nmol MDA g ⁻¹ f.w.)at spike				
Variety (V)↓		till	ering stage			initiation stages				
Treatment (T) \rightarrow		Sulfur ae	rosols (30kg	g ha ⁻¹)		Sulfur aerosols (30kg ha ⁻¹)				
	CaSO ₄	(NH4)2SO4	K ₂ SO ₄	Control	Mean	CaSO ₄	(NH4)2SO4	K ₂ SO ₄	Control	Mean
GW-322	3.967	1.153	3.595	3.105	2.955	3.767	3.96	3.283	3.288	3.574
GW-366	3.54	3.154	1.739	2.18	2.653	3.912	3.992	3.805	3.239	3.737
GW-273	2.895	3.418	3.69	3.501	3.376	4.185	3.594	3.507	3.473	3.689
GW- 173	2.175	2.701	2.262	2.297	2.359	4.725	3.96	2.882	3.443	3.752
JW-336	2.305	2.68	2.485	1.16	2.157	3.949	3.278	4.825	3.319	3.843
Mean	2.976	2.621	2.754	2.449		4.108	3.757	3.66	3.352	
Factor		CD (0.05)	S.Ed(±)			CD (0.05)	S.Ed(±)			
Factor (V)		n.s.	0.442			n.s.	0.285			
Factor (T)		n.s.	0.395			n.s.	0.255			
Interaction (Tx	(V)	n.s.	0.884			n.s.	0.57			
n.s. Non signifi										

Table 6: Effect of sulphur aerosols on Cell Membrane Stability (CMS) of the wheat crop at maximum tillering and spike initiation stages

Variaty (V)	CM	S at the maxi	mum tilleri	ng stage		CMS at spike initiation stage					
$\frac{v \text{ ariety } (v)}{Treatment (T)}$		Sulfur aeros	ols (30kg h	a ⁻¹)		Sulfur aerosols (30kg ha ⁻¹)					
freatment (1)→	CaSO ₄	(NH4)2SO4	K ₂ SO ₄	Control	Mean	CaSO ₄	(NH4)2SO4	K ₂ SO ₄	Control	Mean	
GW-322	0.133	0.019	0.026	0.018	0.049	0.026	0.022	0.022	0.013	0.021	
GW-366	0.022	0.021	0.024	0.13	0.049	0.022	0.024	0.022	0.016	0.021	
GW-273	0.023	0.02	0.027	0.019	0.022	0.026	0.026	0.02	0.015	0.021	
GW- 173	0.125	0.02	0.021	0.021	0.047	0.022	0.024	0.019	0.016	0.02	
JW-336	0.118	0.017	0.019	0.018	0.043	0.019	0.018	0.017	0.015	0.017	
Mean	0.084	0.019	0.023	0.041		0.023	0.022	0.02	0.015		
Factor	r	CD (0.05)	S.Ed(±)				S.Ed(±)				
Factor (V)		n.s.	0.033			n.s.	0.072				
Factor (T)		n.s.	0.03			n.s.	0.064				
Interaction (TxV) n.s. Non Significant		n.s.	0.066			n.s.	0.143				

Table 7: Effect of sulfur aerosols on sulphur and Nitrogen contents in grains at harvest of the wheat crop

Variater (V)	(A) Sulfı	ır content (%) in grain a	t harvest		(B) Nitro	ogen content	(%) in gr	ain at haı	vest		
variety(v)	S	ulfur aerosol	s (30 kg ha ⁻	¹)			Sulfur aerosols (30 kg ha ⁻¹)					
Treatment (1)	CaSO ₄	CaSO ₄	(NH4)2SO4	K ₂ SO ₄	Control	Mean	(NH4)2SO4	K ₂ SO ₄	Control	Mean		
GW-322	0.346	0.153	0.332	0.275	0.308	0.267	0.268	0.292	0.287	0.27		
GW-366	0.276	0.214	0.357	0.156	0.046	0.193	0.288	0.214	0.271	0.256		
GW-273	0.34	0.266	0.332	0.193	0.163	0.238	0.291	0.3	0.195	0.277		
GW- 173	0.295	0.282	0.362	0.18	0.347	0.293	0.318	0.272	0.272	0.285		
JW-336	0.326	0.315	0.32	0.417	0.35	0.35	0.293	0.255	0.223	0.26		
Mean	0.316	0.246	0.34	0.244	0.243		0.291	0.267	0.249			
Factor		CD (0.05)	S.Ed(±)			CD (0.05)	S.Ed(±)					
Factor (V)		n.s.	0.025			0.093	0.044					
Factor (T)		n.s.	0.022			n.s.	0.039					
Interaction (TxV) n.s. Non Significant		n.s.	0.049			n.s.	0.088					

Table 8: Effect of sulfur aerosols on Sulfur Use efficiency (SUE) Nitrogen Use Efficiecy (NUE) in grains at harvest of the wheat crop

Variate (V)	(A	A) SUE (%) i	n grain at l	narvest		(B) NUE (%) in grain at harvest					
$variety(v) \downarrow$ Treatment (T)		Sulfur aeros	sols (30 kg l	ha ⁻¹)		Sulfur aerosols (30 kg ha ⁻¹)					
$1 \text{ reatment} (1) \rightarrow$	CaSO ₄	(NH ₄) ₂ SO ₄	K ₂ SO ₄	Control	Mean	CaSO ₄	$(NH_4)_2SO_4$	K ₂ SO ₄	Control	Mean	
GW-322	0.193	0.227	0.188	0.147	0.189	0.18	0.357	0.205	0.128	0.217	
GW-366	0.198	0.306	0.204	0.086	0.199	0.15	0.273	0.264	0.176	0.216	
GW-273	0.164	0.202	0.194	0.153	0.178	0.263	0.296	0.19	0.097	0.211	
GW- 173	0.199	0.215	0.134	0.141	0.172	0.131	0.203	0.111	0.148	0.148	
JW-336	0.172	0.201	0.13	0.126	0.157	0.114	0.135	0.092	0.157	0.125	
Mean	0.185	0.23	0.17	0.131		0.168	0.253	0.172	0.141		
Factor		CD (0.05)	$S.Ed(\pm)$			CD (0.05)	S.Ed(±)				
Factor (V)		n.s.	0.046			0.068	0.032				
Factor (T)		0.086	0.041			0.061	0.029				
Interaction (TxV) n.s.: N	on Significant	n.s.	0.091			n.s.	0.064				

Table 9: Effect of Sulfur aerosols on Calcium content of the whea	at crop	at spike	initiation s	stage
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Variaty (V)	Exc	changeable [Ca	a ²⁺] (mgg ⁻¹ d	l. w.)		Intercellular [Ca ²⁺] (mgg ⁻¹ d.w.)					
$v \text{ ariety } (v) \downarrow$ Treatment (T)		Sulfur aerosol	s(30 kg ha ⁻¹	l)		Sulfur aerosols(30 kg ha ⁻¹)					
Treatment $(1) \rightarrow$	CaSO ₄	(NH4)2SO4	K ₂ SO ₄	Control	Mean	CaSO ₄	(NH4)2SO4	K ₂ SO ₄	Control	Mean	
GW-322	1.895	1.43	1.295	1.825	1.611	1.535	1.335	1.385	0.56	1.204	
GW-366	1.74	1.74	1.14	1.19	1.453	1.86	1.33	1.455	1.065	1.428	
GW-273	1.125	1.33	1.175	1.22	1.213	1.645	1.345	1.275	0.63	1.224	
GW- 173	1.325	1.135	1.21	1.145	1.204	1.225	1.29	1.04	1.34	1.224	
JW-336	1.34	1.14	1.265	1.32	1.266	1.29	1.495	1.205	1.13	1.28	
Mean	1.485	1.355	1.217	1.34		1.511	1.359	1.272	0.945		
Factor		CD (0.05)	S.Ed(±)			CD (0.05)	S.Ed(±)				
Factor (V)		n.s.	0.16			n.s.	0.153				
Factor (T)		n.s.	0.143			0.288	0.137				
Interaction (TxV) n.s. N	on Significant	n.s.	0.32			n.s.	0.306				

Table 10: Effect of Sulphur aerosols on Potassium content of the wheat crop at spike initiation stage

Variate (V)		Exchangeable	Intercellular [K ⁺](mgg ⁻¹ d.w.)							
$ v arrely (v) \downarrow $		Sulfur aeros	ols(30 kg ha	i ⁻¹)		Sulfur aerosols(30 kg ha ⁻¹)				
Treatment $(1) \rightarrow$	CaSO ₄	(NH4)2SO4	K ₂ SO ₄	Control	Mean	CaSO ₄	(NH4)2SO4	K ₂ SO ₄	Control	Mean
GW-322	1.475	1.34	1.625	1.39	1.458	1.465	1.295	1.22	1.245	1.306
GW-366	2.22	2.38	2.425	1.29	2.079	1.795	1.66	2.265	1.915	1.909
GW-273	1.155	1.94	1.635	1.82	1.638	1.96	1.84	1.435	1.34	1.644
GW- 173	1.855	1.194	2.35	1.767	1.791	1.705	1.57	1.445	1.275	1.499
JW-336	1.88	1.79	1.74	1.325	1.684	1.29	1.39	2.34	1.43	1.613
Mean	1.717	1.729	1.955	1.518		1.643	1.551	1.741	1.441	
Factor		CD (0.05)	S.Ed(±)			CD (0.05)	S.Ed(±)			
Factor (V))		n.s.	0.197			n.s.	0.196			
Factor (T)		n.s.	0.176			n.s.	0.175			
Interaction (TxV) n.s. Non Significant		n.s.	0.394			n.s.	0.392			

Discussion

The main objective behind the recent inquisition was to look into the responses of wheat genotypes to misting of S-aerosols at two prominent spike initiations and heading stages of wheat crop. It unfurled a few natural features of importance, and exhilarated reality on the physiological traits of the wheat crop that are influenced by the Sulfur aerosols. It was, too, attempted to delve down the mechanisms of changing and resetting of cell membrane stability of wheat crop exposed to the S-aerosols.

In the investigation, rate of Pⁿ increased significantly due to Sulfur aerosols. Among the aerosols, the highest increment was in $(NH_4)_2SO_4$ (20.2%) followed by K₂SO₄ (18.3%), and the lowest was in CaSO₄ (16.3%) as compared to control at maximum tillering stage. At spike initiation stage also, rate of Pⁿ increased significantly due to the S-aerosols. Among the aerosols, K₂SO₄ (12.4%) increased maximum rate of Pⁿ followed by CaSO₄ (10.4%) & $(NH_4)_2SO_4$ (10.4%) as compared to control. The rate of Pⁿ in wheat varieties varied significantly following foliar feeding. The Pⁿ rate in crop varieties also differed due to the differences in nitrogen input from the aerosols at their growth stages. In the study, the Saerosols were applied at two different growth stages of wheat crop. The Pⁿ rate in the crop varieties also differed due to the differences in nitrogen input from the aerosols at their growth stages. The Pⁿ depression in higher plants by nitrogenous pollutants was reported previously by Hill and Bennett (1970) ^[11]; Bharali and Bates (2015, 2016) ^[4, 5]. One of the reasons for decreases in Pⁿ rate was the presence of higher chloroplastic pH due to outnumbering protons in the chloroplast than required (six) for a NO₂ reduction. Such build up in pH is prone to damaging of ribulose-1-5- bisphosphate carboxylase/oxygenase. The secondary influence on the ultrastructure of the chloroplast in plants caused by NO₂ might be linked to Pⁿ depression (Heldt *et al.*, 1986) ^[10].

In the present study, NR activity increased significantly due to Sulfur aerosols. Among the aerosols, the highest increment was in $(NH_4)_2SO_4$ (24.5%) followed byK₂SO₄ (22.3%), and the lowest was in CaSO₄ (20.7%) as compared to control at maximum tillering stage. At spike initiation stage also, NR activity increased significantly due to the aerosols. Among the aerosols, $(NH_4)_2SO_4$ (7.1%) increased maximum NR activity

followed by K_2SO_4 (6.6%), and the lowest was in CaSO₄ (3.9%) as compared to control. It was noted that the NR activity in wheat was lower in shoots at the different stages of the crop. The wheat varieties also differed in respect of NR activity in their leaves. It is a fact that NR enzyme catalyses the reduction of nitrate to nitrite, and its levels of activity are determined by the supply of nitrate. The inhibition of NR activity might be due to the feed-back inhibition of the larger ammonium ions (Hisamatsu *et al.*, 1988)^[12].

In the experiment, chlorophyll content in leaf tissue increased significantly due to Sulfur aerosols. Among the aerosols, the highest increment was in (NH₄)₂SO₄ (24.6%) followed by K_2SO_4 (18%), and the lowest was in CaSO₄ (10.1%) as compared to control at maximum tillering stage. At spike initiation stage also, chlorophyll content in leaf tissue increased significantly due to the aerosols. Among the aerosols, K₂SO₄ (23%) increased the maximum chlorophyll content in leaf tissue followed by CaSO₄ (20.4%), and the lowest was in (NH₄)₂SO₄ (0.63%) as compared to the controlled plants. Nitrogen impacts on structure of chlorophyll and protein molecules (Bojovič & Markovič, 2009) ^[6]. It was reported that foliar feeding of plants with amino acids, microelements and Sulfur increased the productivity of spring barley (Staugaitis, Petrauskienė, 2006 ^[22]; Pranckietienė et al., 2015 ^[16]) due to cholorophyll enhancement in leaves.

In the study, NUE increased significantly due to the Saerosols. Among the aerosols, the highest increment in NUE was shown by $(NH_4)_2SO_4$ (44.2%) followed by K_2SO_4 (18%), and the lowest was recorded by CaSO₄ (16%) as compared to controlled one. The S-aerosols increased NUE in the varieties, and the varieties showed remarkable differences in their NUE in grains. However, NUE in grains decreased with the higher concentration of nitrogen aerosols in the experiment. Cations viz., [Ca2+] and [K+] present in intercellular and exchangeable sites varied non-significantly among the varieties due to the S-aerosol treatments. The S-aerosols viz., (NH₄)₂SO₄, CaSO₄, K₂SO₄ couldn't deplete cations substantially from the water free spaces and exchangeable sites of the cell as compared to controlled plants, because the aerosols were in the same range of pH (5.34 to 5.66), where it was neither acidic nor very basic in nature. The S-aerosols exerted beneficial effects on most of the physiological parameters including yield and yield attributes. There were no significant changes in CMS of the wheat varieties as no significant impacts of the S-aerosols were found on either the lipid peroxidation or leaching of cations from the cellular locations. Rather, the aerosols increased N and S contents and their use efficiencies (i.e. NUE & SUE) in grains at harvest. The S-aerosols were not involved in damaging the membrane, Therefore, no loosening or breaking of the membrane to leach out the ions which were plausibly adhered to the intercellular and exchangeable sites of the cells. Hence, whatever intercellular and exchangeable ions were present; they could be recovered in the extracting solutions following incubations with distilled water and SrCl₂ respectively. Moreover, the S-aerosols could increase the [Ca²⁺] and [K⁺] ions in the cellular locations following the treatments of wheat crop with respective aerosols. Thus, in case of Ca_2SO_4 , intercellular [Ca²⁺] (by 0.54-9.7%) and exchangeable [Ca²⁺] (by 4.8-21.1%) increased as compared to the control. Likewise, the intercellular $[k^+]$ (by 34-46.9%) and exchangeable $[k^+]$ (by 21.8-27.7%) enriched in the cellular locations following the K₂SO₄ aerosol treatment. In plant cells, calcium plays integral role on fabricating plasma membrane, and supports in sustaining membrane stability

(Legge *et al.*, 1982) ^[16]. Calcium ions characteristically combine with modulator proteins e.g. Calmodulin (CaM) and it forms Ca-CaM (Dieter, 1984) ^[8] to display sub-clinical signaling that brings about physiological differences and productivity of the crop.

In general, ammonia (gaseous NH₃ or particulate NH₄⁺) or so called NHy (Bharali *et al.*, 2017) ^[3] at high concentration (>1 Mm) is toxic to plants (Mehree and Mohr, 1989) ^[18]. Apart from disconnecting electron transport chain in chloroplast (Lilley *et al.*, 1975) ^[17], ammonia lessens cations viz., $[Ca^{2+}]$, $[K^+]$ and $[Mg^{2+}]$ (Boxman. *et al.*, 1991) ^[7]. In contrast, in the present investigation, perhaps, (NH₄)₂SO₄ on dissociation produced NH₄⁺ and SO₄⁺ where, SO₄⁺ neutralized the cellular acidity produced by NH₄⁺, and it was channeled to participate in N-assimilation pathway to increase its use efficiency in the grain. Because, in aqueous solution, SO₂ exists in three ionic forms viz. Sulphite (SO₃²⁻) leading to SO₄⁺ at pH>6.0, Bisulphite (HSO₃⁻) at pH 2-4 and un-dissociated sulphurous acid (H₂SO₃) dominates at pH 1 (Sounders and Wood, 1973) ^[23].

In the present work, GW-366 followed by 273 emerged as the most efficient genotype physiologically, as these possessed higher NUE, SUE, CMS at crop maturity. The genotype GW-366, also contained higher cellular ions viz., $[Ca^{2+}]$ and [K+] in the intercellular and exchangeable sites. Further, as regard to the field application of sulfur aerosols, $(NH_4)_2SO_4$ was more effective than CaSO₄ and >K₂SO₄ and controlled distilled water. Thus, among the aerosols, $(NH_4)_2SO_4$ @30kgha⁻¹ could be applied as foliar spray to explore the potential economic yield of the selected wheat varieties. Because, $(NH_4)_2SO_4$ not only supported S and N-nutrition, but also maintained the CMS with higher retention of intercellular and exchangeable ions in wheat.

From the forgoing dialogue, it is evident that S and N derived from the aerosol (NH₄)₂SO₄ might have played pivotal roles in physiological process of wheat crop including the quality and quantity of wheat storage protein. N is one of the important constituents of nucleotides, proteins, chlorophyll and enzymes. Similarly, sulfur is involved in plant growth, metabolism and enzymatic reactions. It's known that sulfur is one of the constituents of amino acids such as cystine, cysteine, and methionine. However, identification of molecular markers e.g. QTL for physiological traits for NUE & SUE, amino acids and fatty acids in wheat varieties (e.g. 'GW-273) following incubation in S-aerosols remains as one of the important thrusts in wheat improvement program in near future.

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