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## Isolation and identification of Anthraquinones from the roots of *Morinda Morindoides* (Rubiaceae)

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**Abstract**

*Morinda Morindoides* is a widely used plant in the traditional Ivorian pharmacopoeia, for the treatment of many microbial infections such as those due to amoeba, fungi, and is also used to treat diarrhea and rheumatic pain. The chemical composition of this species is very varied and includes numerous anthraquinones. In this article, we describe the isolation of a new Anthraquinone, 4-hydroxy-1-methoxy-2-methylanthraquinone (1) *M Morindoides* in conjunction with the known anthraquinones (2), 2-methylanthraquinone (3), Soranjidiol (4), Damnacanthol (5), Damnacanthal (6), Rubiadin-1-methyl-ether (7). The structures were elucidated through spectroscopic studies including 2D-NMR experiments (HSQC, HMBC, COSY, and NOESY).

**Keywords:** *Morinda Morindoides*, Rubiaceae, 4-hydroxy-1-methoxy-2-methylanthraquinone, anthraquinones

**Introduction**

*Morinda Morindoides* is a very popular medicinal plant in many African countries. This plant is used against diarrhoea, amoebiasis, rheumatic pains and fungi [1, 2]. In Ivory Coast, the aqueous decoctions of leaves or roots are widely used for the treatment of malaria [3]. Many traditional uses were confirmed by reports on biological studies [4-8]. The ethyl acetate extract showed significant plasmodial and antidiarrheal activity [6, 9, 10]. In this present paper, we report the isolation and identification of the Anthraquinone, 4-hydroxy-1-methoxy-2-methylanthraquinone (1) from a natural source which was identical to a known synthetic product [11, 12].

**Experimental**

**General:** Melting points were determined with a Büchi B-545 and were uncorrected. The optical rotation was measured on a Schmidt-Haensch Polartronic HH8 polar meter. The UV spectra were obtained by using a Philips PU 8720 spectrophotometer. The IR spectra were recorded on a Bruker Vector 22 FT-IR spectrometer. The EIMS were recorded on Varian MAT-312 mass spectrometer. The HREIMS were measured on a Micro mass Q-TOF micro instrument (Manchester, UK). <sup>1</sup>H NMR (400 MHz) and <sup>13</sup>C NMR (100 MHz) spectra were recorded in DMSO-d<sub>6</sub> on a Bruker Avance DRX-400 spectrometer with TMS as internal standard. Column chromatography and gel permeation were run on Merck silica gel 60 and Sephadex LH 20. Analytical TLC was carried out on 0.25 mm thick layer of silica gel percolated on aluminium foil (Merck GF254). Spots on chromatograms were detected by observing under UV light (254 nm) and were further visualized by spraying with a vanillin-H<sub>2</sub>SO<sub>4</sub>.

**Plant material:** Roots of *Morinda Morindoides* were collected in July 2009 in Saioua, in the West of Ivory Coast. The plant was identified by Prof. Aké Assi of the University of Cocody-Abidjan. A voucher specimen (ZG N° 116) was deposited at the "Centre National de Floristique" of the University of Cocody-Abidjan. The collected plant materials were dried and the dry samples were crushed and stored at 10°C until use.

**Extraction and isolation:** Dried and powdered roots of *Morinda Morindoides* were extracted three times with 80% aqueous ethanol. After filtration, the combined extracts were concentrated at 40°C in the rotary evaporator under reduced pressure. The resulting residue was suspended in water and successively partitioned with petroleum ether, ethyl acetate and n-butanol. The ethyl acetate extract (15.23 g) was subjected to silica gel column chromatography

with a gradient of AcOEt in Et<sub>2</sub>O (from 100:0 to 0:100). Five fractions (F<sub>1</sub> to F<sub>5</sub>) were obtained were these dried in the rotary evaporator. The fraction F<sub>3</sub> contained yellow crystals; their recrystallization from Et<sub>2</sub>O gave the compound 1 (70 mg). The fraction F<sub>4</sub> was repeatedly separated on Sephadex LH-20 columns elated with CHCl<sub>3</sub>/MeOH (1:1) to yield three compounds 2 (24 mg), 3 (47 mg) and 4 (42 mg). The fraction F<sub>5</sub> was eluted repeatedly on Sephadex LH-20 chromatography column with CHCl<sub>3</sub>/MeOH (7:3) to yield three compounds 5 (47 mg), 6 (39 mg) and 7 (45 mg).

**4-hydroxy-1-methoxy-2-methylanthraquinone (1):** Yellow amorphous powder (70 mg); MP 176-178°C; UV (MeOH): λ<sub>max</sub> (nm) (log ε) 202 (1.04), 278 (1.35); IR(KBr): ν<sub>max</sub> cm<sup>-1</sup>: 3308, 1671, 1565, 1446, 1414, 1335, 1300; HR-ESI-MS: m/z [M+Na]<sup>+</sup> 291.0630 (calcd for C<sub>16</sub>H<sub>12</sub>O<sub>4</sub>Na: 291.0633); <sup>1</sup>H-NMR, <sup>13</sup>C-NMR (see Table 1).

### Results and discussion

Chromatographic separations on silica gel of the ethyl acetate extract of the roots of *Morinda Morindoides* have led to the isolation of Anthraquinone 1. Six other compounds were isolated: Rubiadin (2) [13, 14], 2-methylanthraquinone (3) [15, 16], Soranjidiol (4) [13, 14], Damnacanthol (5) [14], Damnacanthal (6) [13, 14, 17] and rubiadin-1-methyl-ether (7) [13, 14]. The structure of compound 1 is presented in Figure 1.

Compound 1 was obtained as a yellow amorphous powder. Its UV and IR spectra were typical of those of anthraquinones [18-20]. The positive ion mode HRESIMS gave a pseudo molecular ion peak at m/z 291.0633 [M+Na]<sup>+</sup>, consistent with

a molecular formula C<sub>16</sub>H<sub>12</sub>O<sub>4</sub>Na. The NMR spectra data supported the presence of Anthraquinone skeleton. The <sup>1</sup>H NMR spectra gave a singlet at δ<sub>H</sub> 13.10 ppm assigned to a chelated hydroxy proton at C-4. Five aromatic protons exhibited as two doublets of doublet at δ<sub>H</sub> 8.25 (1H, J = 7.5 and 1.5 Hz, H-8) and δ<sub>H</sub> 8.16 (1H, J = 7.5 and 1.5 Hz; H-5), two triplets of doublet at δ<sub>H</sub> 7.84 (1H, J = 7.5 and 1.5 Hz; H-7) and δ<sub>H</sub> 7.76 (1H, J = 7.5 and 1.5 Hz; H-6) and a singlet at δ<sub>H</sub> 7.20 ppm (H-2). These five protons suggested that A ring was unsubstituted and B ring trisubstituted. A Methoxy group appeared as a singlet at δ<sub>H</sub> 3.86 ppm and a methyl group as a singlet at δ<sub>H</sub> 2.42 ppm. The <sup>13</sup>C NMR spectrum showed 16 carbon signals: 12 aromatic carbons, one methoxy (δ<sub>C</sub> 61.1 ppm), one methyl (δ<sub>C</sub> 16.5 ppm), and two carbonyls (δ<sub>C</sub> 183.5 and 182.1 ppm) characteristic of anthraquinones. In the HMBC spectrum (Figure 2), the signal of the hydroxyl proton (δ<sub>H</sub> 13.10 ppm) correlated with those of C-3, C-4 and C-4a. The cross peaks between the signal of the methyl protons group and those of C-1 (δ<sub>C</sub> 153.5), C-2 (δ<sub>C</sub> 127.6) and C-3 (δ<sub>C</sub> 116.1) afforded the positions of the methyl and the methoxyl groups at C-2 and C-1. The assignments of C-3, C-5, C-6, C-7 and C-8 were determined from the HSQC correlations. The structure of the compound 1 was identified as 4-hydroxy-1-methoxy-2-methyl Anthraquinone (named morindin). It was identical to a synthetic Anthraquinone obtained by the reaction of phthalide anions with Quinone monoacetals [11] and the reaction of 3-cyano-1(3*H*)-isobenzofuranone with *N*-methyl-*p*-methoxy semi Quinone animal [12].

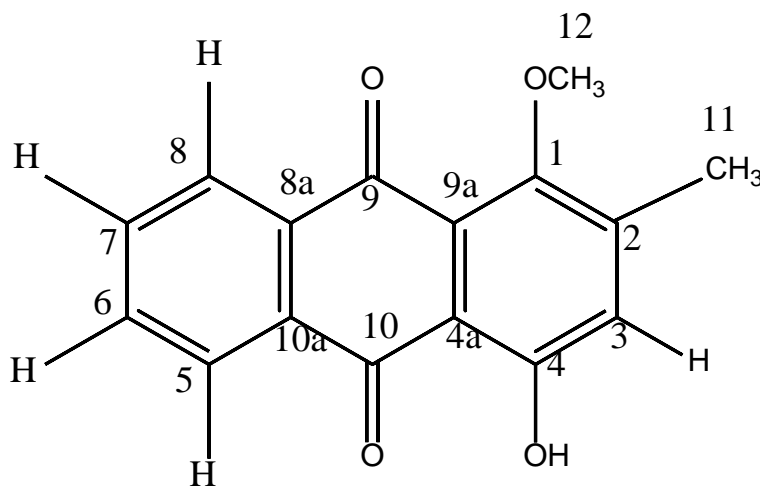


Fig 1: Chemical Structure of compound 1

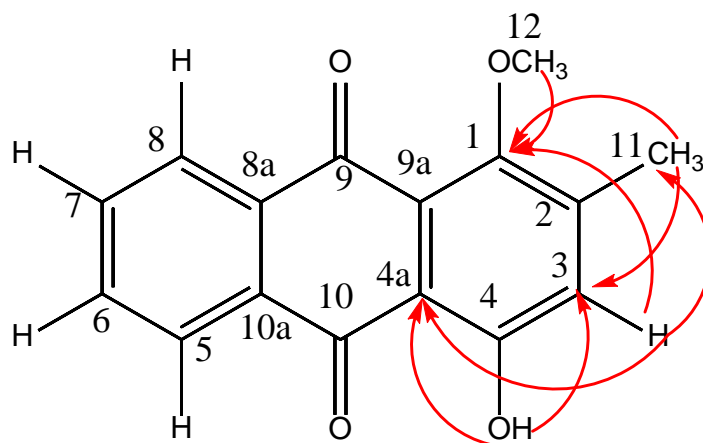


Fig 2: Structure and Key HMBC (H → C) correlations of compound 1

**Table 1:** NMR spectral data of compound 1 in DMSO-d<sub>6</sub>

Position	$\delta_C$	$\delta_H$ [(m, J (Hz))]	COSY	HMBC <sup>b</sup>
1	153.48 (q) <sup>a</sup>	-		
2	127.63 (q)	-		
3	116.03 (t)	7.20 (s)		1, 2, 4, 4a, 11
4	160.08 (q)	-		
4a	132.20 (q)	-		
5	123.82 (t)	8.16 (dd:7.5;1.5)	H-6	6, 7, 8a, 10, 10a
6	133.81 (t)	7.76 (td:7.5;1.5)	H-5, H-7	5, 7, 8, 10a
7	134.78 (t)	7.84 (td:7.5;1.5)	H-6, H-8	5, 6, 8, 8a
8	126.52 (t)	8.25 (dd:7.5;1.5)	H-7	6, 7, 8a, 9, 10a
8a	145.96 (q)	-		
9	182.16 (q)	-		
9a	127.02 (q)	-		
10	183.45 (q)	-		
10a	135.14 (q)	-		
11	16.48 (p)	2.42 (s)		1, 2, 3
12	61.07 (p)	3.86 (s)		1
4-OH	-	13.10 (s)		3, 4, 4a

From a chemotaxonomic viewpoint, similar 1, 2, 4-trisubstituted B-ring anthraquinones, such as 1 were obtained in *Prismatomeris Connata* [21, 22], *Digitilis lanata* [23] and *Tectona grandis* [24] belonging respectively to Rubiaceae, Scrophulariaceae and Verbenaca family. Rubiadin (2) was previously isolated from the roots of *M. citrifolia* [14], *M. angustifolia* [25] and *M. Elliptica* [26]. 2-methylanthraquinone (3) was isolated from the roots of *M. umbellata* [27] and *M. officinalis* [28]. Soranjidiol (4) was isolated from the roots of *M. citrifolia* [14] and *M. elliptica* [26]. Damnacanthol (5) was identified from the roots of *M. umbellata* [27] and *citrifolia* [14]. Damnacanthol (6) was obtained from the roots of *M. citrifolia* [14] and *M. elliptica* [26]. Rubiadin-1-methyl-ether (7) was isolated from roots and stems of *M. umbellata* [27], *M. citrifolia* [14] and *M. elliptica* [24]. This is the first report of the occurrence of morindin (1) as a new natural Anthraquinone, though it has artificially prepared from 3-cyano-1(3H)-isobenzofuranone and N-methyl-p-methoxy semi quinone animal [12]. They are to our knowledge describe here for the first time in *M. Morindoides* roots. Therefore, they could be used to establish a relationship between these species.

## Conclusion

The results of this study, obtained for the first time from *Morinda Morindoides* roots, led to the isolation and characterization of seven anthraquinones (morindin, rubiadin, 2-methylanthraquinone, Soranjidiol, Damnacanthol, Damnacanthol and rubiadin-1-methyl-ether). The complete attribution of the structures was established by inspection of NMR spectral data (<sup>1</sup>H and <sup>13</sup>C) and by mass spectrometry. However, one compound, morindin was isolated for the first time from a natural source. This compound has a hydroxyl group attached at 4-position, a methoxy group attached at the 1-position and a methyl group attached at the 2-position of Anthraquinone. Anthraquinones are well known in the literature for their broad biological therapeutic. The results of our study could explain the biological activity of *Morinda Morindoides* Roots.

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